GUAIANOLIDES FROM LIATRIS SQUARROSA*

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Key Word Index—Liatris squarrosa; Compositae; Eupatorieae; sesquiterpene lactones; guaianolides, spicatin analogs.

Abstract—The isolation of three new guaianolides related to spicatin from Liatris squarrosa is reported.

INTRODUCTION

In continuation of our studies of *Liatris* species (Compositae, Eupatorieae) which have resulted in the isolation of a number of cytotoxic and antileukemic sesquiterpene lactones [1-12] we have investigated *Liatris squarrosa* (L.) Michx. and have isolated three new guaianolides, desacetylspicatin (1a), desacetylspicatin hydrochloride (2a) and 2b. Sitosterol, β -amyrin and lupeol were also found.

RESULTS AND DISCUSSION

¹H and ¹³C NMR spectra of the three new sesquiterpene lactones, all of which were non-crystalline, are given in Tables 1 and 2. In each case, spin-decoupling experiments established the sequence shown in partial structure A (numbering as in final structure), with H-3 being allylically coupled to the methyl group on C-4. In the case of 1a, the empirical formula from the MS and the spectroscopic evidence also showed the presence of partial structure B which on biogenetic grounds was inserted between C-1 and C-9. The spectrometric evidence (MS, Tables I and 2) showed clearly that the acyl function attached to C-8 of all three compounds was the *cis*-sarracenoyl-sarracenate ester moiety first encountered in spicatin (1b) and its congeners. [5, 6].

On this basis and because of the coupling constants, la was deduced to be desacetylspicatin as can be seen by comparing its ¹H and ¹³C NMR spectra with those of spicatin (1b) whose structure has been settled by X-ray crystallography [5,6]. The only significant differences in the ¹³C NMR spectra are the expected upfield shift of C-2 and the downfield shift of C-3 in going from 1b to 1a.

Confirmatory evidence was obtained by acetylation of 1a and 1b which gave the same diacetate 1c.

MS and NMR evidence (Tables 1 and 2) indicated that the second lactone was the desacetyl analog 2a of spicatin hydrochloride (2c). This was confirmed by converting 1a to 2a. Although 2a could conceivably be an artefact, a number of similar compounds have been reported as natural products.

The third lactone was a cis-sarracenovl analog of 2a (MS), in which the new sarracenoyl group esterified a primary hydroxyl as in 2b or 2d. A decision in favor of 2b was reached as follows. Close inspection of the ¹³C NMR spectra of 1a, 1b and 2a indicated that the signals of the terminal and non-terminal cis-sarracenoyl ester moieties differ significantly not so much in the chemical shifts of C-5' and C-5", which are surprisingly similar, but in the chemical shifts of C-2' and C-2", the C-2 carbon of the sarracenoyl group experiencing an upfield shift of ca 4 ppm, from ~ 131.5 to 127.5, on acylation. This is also evident on comparing the ¹³C NMR spectra of 3a and 3b [13]. As the ¹³CNMR spectrum of the third lactone exhibited only one singlet near 127.5 and two singlets near 131.50 ppm, it must contain two terminal cis-sarracencyl ester functions as in 2b. A formula in which the attachment of the C₅ and C₁₀ ester functions to C-8 and C-14 is reversed is not excluded by the NMR spectra, but seems less likely for reasons of analogy; an attempt at selective hydrolysis to verify this failed.

Our results differ from those of a previous study of *L. squarrosa* which recorded the absence of sesquiterpene lactones [14]. *L. squarrosa* has been segregated into several varieties [15]. Our material was difficult to classify but may well be var. *squarrosa*. The sesquiterpene lactone chemistry

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Table 1. ¹H NMR spectra of compounds from Liatris squarrosa*

	1a	1a†	1b	1c	2a	2a†	2b
H-1	2.0‡	1.42‡	1.63 dd	1.61	2.42‡	2.27	2.36 dd
H-2	4.62 ddq	4.20*	5.44	5.44	5.46	4.42	4.52
H-3	5.70 dq	5.45	5.48	5.50	5.75	5.57	5.75
H-5	2.76 dd	2.21	2.18	2.22	2.75	2.45	2.76
H-6	4.64 dd	4.63	4.61	4.61	4.50	4.65	4.57
H-7	3.56 dddd	3.40	3.38	3.40	3.90	3.87	3.95
H-8	5.62 ddd	5.77	5.72	5.76	5.70	5.88	5.71
H-9a	2.65 dd	2.75	2.33	2.37	2.42	2.60	2.40
H-9b	2.22 dd	2.13	2.05	2.06	2.42	2.40	2.40
H-13a	6.29d	6.24	6.25	6.24	6.25	6.28	6.24
H-13b	5.56 d	5.18	5.20	5.24	5.47	5.33	5.46
H-14	2.57 §	2.248	2.22§	2.25§	3.65 d	3.46	4.38§
					3.90 d	3.79	· ·
H-15	1.99 dd	1.89	1.84	1.87	2.02 dq	2.01	2.02
H-3'	7.10 q	7.04	6.94	6.94	7.14	7.11	7.12
H-4'	1.95	1.35	1.39	1.40	1.95	1.47	1.96
H-5'	4.89 d	4.80	4.87	4.90	4.93	4.84	4.90
	4.80 d	4.59	4.65	4.70	4.76	4.72	4.80
H-3"	6.91 q	6.80	6.83	6.99	6.91	6.95	6.91
H-4"	$1.90\dot{d}$	1.48	1.5	1.58	1.90	1.59	1.91
H-5"	4.30§	4.248	4.31 §	4.89§	4.30§	4.29 §	4.31 §
Misc	v	Ū	9	٩ "	Ü	Ü	¶″

 $J_{1,2} = 7$ Hz for 1-3, 6 Hz for 4 and 5; $J_{1,5} = 9$ Hz for 1-3, 6 Hz for 4 and 5; $J_{2,3} = 2$; $J_{2,15} = 0.5$; $J_{3,15} = 1$; $J_{5,6} = 11$; $J_{6,7} = 8.5$; $J_{7,8} = 3.9$, $J_{3,13a} = 3.5$; $J_{7,13b} = 3$; $J_{8,9a} = 8$; $J_{8,9b} = 7$; $J_{9a,9b} = 15$; $J_{3'',4''} = J_{3'',4''} = 7$; $J_{4a',4b'} = 13$.

 CH_2R

^{*} Run at 270 MHz in CDCl₃ unless indicated otherwise. Frequencies in ppm downfield from TMS.

 $[\]dagger$ In C_6D_6 .

[‡] Obscured signal.

[§]Intensity two protons; centre of AB system.

^{||} Intensity three protons.

In 1b. 1.72 (Ac). In 1c, 1.73 and 1.63 (Ac). In 2b, 7.02 q (7 Hz, H-3"), 1.94 d (7, H-4"), 4.31 br (H-5").

Table 2. 13C NMR spectra of compounds from Liatris squarrosa*

	1a	1b	2a	2b
C-1	55.37 d	55.56 d†	54.66 d	54.77 d
C-2	76.37 d	78.68 d†	74.70 d	74.75 d
C-3	129.02 d	125.34 d†	128.91 d	128.99 d
C-4	149.51	150.59	149.06	149.18
C-5	52.44 d	51.20 d†	51.90	52.29
C-6	81.15	80.21 d†	82.06 d	82.12 d
C-7	47.64 d	47.53 d†	47.22 d	47.30 d
C-8	67.46 d	67.19 d†	67.27 d	67.38 d
C-9	36.15 t	35.68 t	36.21 t	35.99 t
C-10	56.05	55.21	73.56	77.30
C-11	134.21	133.65	134.68	134.80
C-12	169.55	169.88	169.87	169.85
C-13	122.53 t	123.02 t	121.83 t	121.69 t
C-14	56.33 t*	56.59 t‡	54.91 t	71.43 t
C-15	17.43 q	17.46 q	17.97 q	17.97 q
C-1′	165.38	165.33	165.70	165.76
C-2'	127.57	127.60	127.50	127.60
C-3'	145.49 d	145.61 d	145.80 d	145.69 d
C-4′	14.53 q	14.53 q	14.57 q	14.57 q
C-5'	57.38 t	57.52 t	57.46 t	57.46 t
C-1"	166.63	166.68	166.82	166.80
C-2"	131.69	131.90	131.44	131.77‡
C-3''	141.87 d	141.11 d	142.30 d	142.31 d§
C-4''	14.23 q	14.17 q	14.31 q	14.30 q
C-5′′	56.49 t‡	56.18 <i>t</i> ‡	56.16 t	56.05 t
C-1'''		169.12		167.46
C-2'''		21.30 q		131.50‡
C-3'''		_		142.15 d§
C-5'''				56.27 t

^{*} Run in CDCl₃ at 67.9 MHz. Unmarked signals are singlets.

of our collection resembles that of *L. spicata*, *pycnostachya*, *graminiflora* and *tenuifolia* all of which yielded similar guaianolides [5, 6, 10], whereas examination of other *Liatris* species gave heliangolides, [1-4, 7-9, 11, 12]. Further studies are required to establish whether such differences provide useful taxonomic information at the subgeneric level.

EXPERIMENTAL

Aerial parts of *Liatris squarrosa* (L.) Michx., collected by Dr. Norlan Henderson on 18 July 1978 along Missouri Highway No. 13 ca 3 miles north of Lowry City, St. Clair County, Missouri (Henderson 78-11 on deposit in herbarium of University of Missouri–Kansas City), wt 5.6 kg, were extracted with CHCl₃ and worked up as usual [16]. The crude gum (200 g) was adsorbed on 220 g of Si gel (Mallinckrodt 100 mesh) and chromatographed on a 2 kg Si gel column packed with toluene – CHCl₃ (1:). The column was eluted with solvents of increasing polarity, 500 ml fractions being collected and monitored by TLC in the following

order: 1-6, CHCl₃-toluene (1:0); 7-15, CHCl₃; 16-20, CHCl₃-MeOH (99:1); 21-25, CHCl₃-MeOH (97:3) and 26-36 CHCl₃-MeOH (19:1).

Fractions 1–6 contained waxy material which was not studied further. Fractions 7–15 which showed one major spot on TLC were combined and crystallized from CHCl₃–MeOH, yield 20 g, mp 202° which was a 1:1 mixture of lupeol and β -amyrin by direct comparison. Fractions 16–20 were recrystallized from MeOH, yield 5 g, mp 135°, which was identified as sitosterol by direct comparison.

TLC of fractions 21–25 showed the presence of a major constituent which was purified by prep. TLC (EtOAc-hexane, 3:1). The non-crystalline lactone **2a** had IR v_{max} cm⁻¹: 3500 (br), 1760 (γ -lactone), 1720 (sh), 1710 (esters), 1650 (sh), 1640 and 1630 (sh) (double bonds); $[\alpha]_D = 40^\circ$ (c 0.02, CHCl₃); MS (low resolution) m/e: 510 (M⁺), 492 (both extremely weak), 474, 456, 438, 376, 360, 358, 340, 322, 309, 278, 260, 242, 224; 99 (base peak. C₅H₇O₂). The composition of the peak at m/e 474 was shown to consist of M = 2H₂O and M = HCl (weak) by peak matching (Calc. for C₂₅H₂₇O₇Cl, C₂₅H₃₀O₉: 474.1449; 474.1890. Found (MS): 474.1431; 474.1876).

[†] Assignment verified by single frequency off-resonance decoupling.

^{‡ § ||} Interchangeable assignments.

Fractions 26–31, which contained one major constituent, were combined and purified by prep. TLC (CHCl₃–MeOH, 9:1). The product **1a** (100 mg) could not be induced to crystallize and had $[\alpha]_D = 25.5^\circ$ (c-0.03, CHCl₃); IR v_{max} cm⁻¹: 3420 (br, –OH), 1770 (γ -lactone), 1730 (sh) and 1720 (strong esters), 1655 (sh), 1645, 1630 (sh) and 1620 (sh); MS m/e: 474 (M⁺, very weak), 456, 438, 420, 359, 358, 340, 260, 24, 99 (base peak, $C_5H_7O_2$). The molecular ion was not observed.

Acetylation of $50 \, \text{mg} \, 1a \, \text{with} \, \text{Ac}_2 O \, \text{Pyat} \, 0^\circ \, \text{overnight} \, \text{followed}$ by the usual work-up and purification by prep. TLC gave $35 \, \text{mg} \, 1c \, \text{which}$ was identical with acetylspicatin in all respects (TLC, IR, NMR). A solution of $20 \, \text{mg} \, 1a \, \text{in} \, 5 \, \text{ml} \, \text{dry} \, \text{EtOH} \, \text{and} \, 0.2 \, \text{ml} \, 12 \, ^\circ ,_0 \, \text{HCl}$ was stirred at room temp, for $4 \, \text{hr} \, \text{and} \, \text{evapd} \, \text{in} \, \text{vacuo}$. The residue was purified by prep. TLC. The product which was less polar than $1a \, \text{was} \, \text{identical} \, \text{in} \, \text{all} \, \text{respects} \, \text{with} \, 2a \, \text{from} \, \text{the} \, \text{plant}$.

Fractions 32–36 showed the presence of a major constituent which was purified by prep. TLC (EtOAc-hexane, 9:1, two developments). The product **2b** (50 mg) could not be induced to crystallize and had $[\alpha]_D = 30.5^\circ$ (c 0.015, CHCl₃); IR v_{max} cm⁻¹: 3414 (br. –OH), 1765, 1725 (sh) and 1710 (v. strong, esters), 1660 (sh), 1655, 1635 (sh) and 1620 (sh); MS m/e: 572 (M = 18, very weak), 554, 474, 456, 438, 358, 340, 260, 242, 99 (base peak, $C_5H_7O_2$). The molecular ion was not observed.

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